

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
17 July 2003 (17.07.2003)

PCT

(10) International Publication Number
WO 03/057753 A1

(51) International Patent Classification⁷: **C08G 59/50**, 59/54, 59/56, C08L 63/00, 63/02

(21) International Application Number: PCT/US02/41424

(22) International Filing Date:
27 December 2002 (27.12.2002)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
60/346,443 28 December 2001 (28.12.2001) US

(71) Applicant (*for all designated States except US*): **ILLINOIS TOOL WORKS INC.** [US/US]; 3600 West Lake Avenue, Glenview, IL 60025 (US).

(72) Inventor; and

(75) Inventor/Applicant (*for US only*): **SURJAN, Jim** [US/US]; Illinois Tool Works Inc., 5N 479 Farrier Point, St. Charles, IL 60175 (US).

(74) Agents: **POSILLICO, Joseph** et al.; Synnestvedt & Lechner, 1101 Market Street, Suite 2600, Aramark Tower, Philadelphia, PA 19107 (US).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: ADHESIVE OF EPOXY COMPOUND, ALIPHATIC AMINE AND TERTIARY AMINE

(57) Abstract: An adhesive for anchoring materials in or to concrete or masonry exhibits a shorter cure time than previous adhesives and comprises an epoxy compound and a curing agent of at least one aliphatic amine and at least one tertiary amine, optionally with a reactive diluent, which possesses sufficient strength to pass ICBO Heat Creep Test at 110 degrees F and the ICBO Damp Hole Test at 75 degrees F.

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ADHESIVE COMPOSITIONS

Cross Reference to Related Application

The present application is related to and claims the priority benefit of United States Provisional Application 60/346,443, which was filed on December 28, 2001 and which is incorporated herein by reference.

Field of the Invention

The present invention relates to epoxy adhesive compositions. More specifically, the invention relates to epoxy adhesive compositions that include a curing agent, also referred to as a hardener, comprising one or more amine compounds. Certain aspects of the invention relate to systems and methods for anchoring materials in or to concrete or masonry.

Background of the Invention

Many applications require that one article, structure or item be bonded or adhered to another article, structure or item. For example, anchor bolts are employed in various fields of engineering as strengthening or reinforcing members in rock formations, or concrete or masonry structural bodies. The bolts, which are typically metallic, are inserted into holes in the rock formations, or concrete or masonry structural bodies, and are fixed or anchored therein by means of an anchor composition. Typically, the anchor composition cures or hardens (polymerizes) to form a strong bond between the rock formation, or the concrete or masonry structural body and the bolt.

It has been known to use curable synthetic resins as the primary adhesive for the safe fastening of anchoring rods, bolts and similar devices in solid rock, masonry, concrete and the like. Typically, but not exclusively, certain of the starting components of the adhesive composition are kept separate from one another and then combined at or near the point of fastening. In such so called two part systems, the components which are kept separate until the time of use are combined at the site and then introduced to the anchor bolt, the bore or hole prepared to receive the bolt, or both. Thus, the formation of the cured

adhesive body that binds the fastener to the base structure is initiated at about the time the fastener is placed in its final position.

In concrete and masonry, anchor bolts are used for reinforcement. Anchor bolts are also used for attaching objects to concrete or masonry. Objects that have been attached to concrete or masonry using anchor bolts include, but are not limited to, electrical conduits, panels, piping and wall sections. Adhesive anchors are preferred over mechanical anchors for anchoring in soft concrete or masonry because adhesive anchors place less stress on the concrete or masonry. As used herein, the term "masonry" shall include, but is not limited to, stone, brick, ceramic tile, cement tile, hollow concrete block and solid concrete block.

A useful anchor composition should provide a strong bond between the concrete or masonry and the material to be anchored to the concrete or masonry, be easy to dispense at the location of use and is preferably not noxious or offensive to the user. It is also preferred that the adhesive is capable of achieving a substantial portion of its ultimate strength in a relatively short period of time. This last feature can be critically important to the commercial success of the adhesive since it can allow construction project to proceed at an efficient and profitable pace. Moreover, applicant has noted that it is desirable in many applications that the adhesive is capable of achieving good bonding properties even if it is used or applied during inclement weather or under other conditions which produce substantial moisture at the bond site.

Epoxy resins are often used as adhesives. Epoxy resins are typically one part of two-part adhesive compositions. The epoxy resin and a curing or hardening agent are typically mixed immediately prior to use, and cure within a certain amount of time. Many curing agents are nucleophilic compounds, such as amines or thiols, with at least two nucleophilic groups. Previously used amine-based curing agents often suffer from the disadvantage of producing adhesives with a relatively long cure time, particularly at relatively low temperatures. While thiol (mercaptan)-based curing agents can sometimes be formulated to react more rapidly, and at lower temperatures, than amine-based curing agents, such compounds generally produce vapors that may be noxious and/or offensive to the user's olfactory senses.

Summary of the Invention

Applicant has come to appreciate a need in the art for adhesive compositions with excellent physical and processing characteristics. In view of applicant's appreciation of this need and the deficiencies in the prior art as described above, it is an object of the present invention to provide an adhesive composition that has excellent heat creep resistance and excellent damp-hole bonding properties.

One embodiment of the present invention achieves these and other objects by providing epoxy-based adhesive compositions, systems and methods that include the use of a hardener comprising amine. The composition preferably has a cure time of about two

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hours or less and achieves sufficient strength to pass ICBO Heat Creep Test (ICBO acceptance criteria AC58) at 110°F. The preferred compositions of the present invention are capable also of passing the ICBO Damp Hole Test at 75°F. In highly preferred embodiments, the present compositions have at once a cure time of about two hours or less, the capability to pass ICBO Heat Creep Test and the capability to pass the ICBO Heat Damp Hole Test. The present compositions and systems also preferably are substantially free of offensive odors, as would be produced by the presence of substantial amounts of mercaptans in the formulation.

In certain preferred embodiments, the present invention provides a composition, preferably for use in systems and methods for anchoring materials in or to concrete or masonry, that comprises a first composition and a second composition that are brought together at or about the time of use. Preferably in such embodiments the first composition comprises a resin component, preferably an epoxy resin, and the second composition includes a hardening or curing agent that comprises amine. In such embodiments, the first and second compositions are brought into intimate contact to form a curable adhesive composition.

As used herein the term "cure time" refers to the time it takes from the initiation of curing reaction for the curable adhesive composition to achieve about 90% of its ultimate load capacity or strength under substantially dry conditions, with ambient temperature at about 70°F.

Although it is contemplated that excellent results can be achieved according to the broad teachings of the present invention as described herein generally, applicant has found that exceptionally surprising and unexpectedly beneficial results are obtainable from adhesive compositions comprising a difunctional liquid epoxy derived from bisphenolA/epichlorohydrin and a curing agent comprising at least one aliphatic amine and at least one tertiary amine. Applicant has found that exceptional results can be achieved in accordance with the present invention by compositions formed from a select class of polymerizable epoxy compounds, preferably in combination with a select reactive diluent for the epoxy and a select curing agent. Furthermore, applicant has discovered that unexpectedly superior performance is possible when the polymerizable epoxy compounds are present in the adhesive compositions in amounts of from about 20 wt% to about 40 wt%, and even more preferably from about 25 wt% to about 35 wt%. Furthermore, applicants have found that unexpectedly superior results are generally achieved when the polymerizable epoxy compounds are present in the adhesive compositions in amounts of from about 70 pbwa to about 95 pbwa, and even more preferably about 85 pbwa to about 95 pbwa, wherein the term "pbwa" means part by weight based on active components. As used herein, the term "active components" refers to those components of the composition which participate in the curing of the composition, either directly as a reactant or indirectly as a catalyst, curing agent, inhibitor and the like. In certain preferred versions of such

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embodiments, the adhesive composition further comprises from about 0% to about 5% by weight of a reactive diluent, which may or may not constitute an epoxy compound in accordance with the present invention.

DETAILED DESCRIPTION

I. THE COMPOSITIONS

As mentioned previously, the preferred curable compositions in accordance with the present invention generally comprise an epoxy compound in reactive combination with a curing agent comprising one or more amine compounds.

The curable composition may be dispensed using various methods known to those skilled in the art. For example, the composition may be dispensed using a dual cartridge device similar to a caulk gun, or the composition may be dispensed using a glass or film capsule. The composition may also be dispensed in bulk from bulk containers using meter-mix equipment, which is known to those skilled in the art. See, for example, U.S. Pat. Nos. 4,651,875, 4,706,806 and 4,729,696, the disclosures of which are hereby incorporated by reference. It is recognized that the amounts of the various components of the composition may vary depending on the type of dispensing system used. In preferred dispensing methods, the curable composition is formed by the mixing of a first composition and a second composition. Typically, the mixing occurs immediately before the curable composition is to be used. For example, when the composition is dispensed using a dispensing gun, the first composition and the second composition, which are contained in separate cartridges of the dispensing gun, may be mixed as they are ejected from the cartridges and applied to the surfaces to be bonded, such as to either the concrete or masonry and/or the anchoring device in the case of anchor compositions. Similarly, when the composition is dispensed using a glass capsule, the capsule is typically comprised of two chambers that contain the first composition and the second composition respectively. When the glass capsule is crushed, the two chambers are crushed and the contents are allowed to mix, forming the anchor composition. When the anchor composition is dispersed in bulk, a first composition and a second composition may be stored in separate bulk containers and combined through pumping with mixing in the appropriate ratio to make the curable composition.

In general, the epoxy-based adhesives of the present invention comprise epoxy and amine-based curing or hardening agent for the epoxy. As used herein, the term "curing agent" refers to one or more components which are capable of catalyzing and/or accelerating the polymerization and/or crosslinking reaction of the resin component, which in preferred embodiments comprises one or more epoxy compounds.

It is contemplated that the relative proportion of epoxy to amine in the curable composition may vary widely within the scope hereof in order to accommodate the needs

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and requirements of any particular application. In general, however, it is preferred that the reactive composition of the present invention have a epoxy:amine weight ratio of from about 0.5:1 to about 10:1, more preferably from about 0.7:1 to about 2:1 and even more preferably from about 0.8:1 to about 1:1.

The Epoxy

As used herein, the term "epoxy compound" refers to a compound or combination of two or more compounds that contain a reactive epoxy group or oxirane ring. Such materials, broadly called epoxides, include monomeric epoxy compounds and epoxides of the polymeric type and can be aliphatic, cycloaliphatic, aromatic or heterocyclic. These materials preferably have, on the average, at least 1.5 polymerizable epoxy groups per molecule, and even more preferably two or more epoxy groups per molecule. The polymeric epoxides include linear polymers having terminal epoxy groups (e.g., a diglycidyl ether of a polyoxyalkylene glycol), polymers having skeletal oxirane units (e.g., polybutadiene polyepoxide), and polymers having pendent epoxy groups (e.g., a glycidyl methacrylate polymer or copolymer). The epoxides may be pure compounds but are generally mixtures containing one, two, or more epoxy groups per molecule. The "average" number of epoxy groups per molecule is determined by dividing the total number of epoxy groups in the epoxy-containing material by the total number of epoxy molecules present.

The epoxy-containing materials may vary from low molecular weight monomeric materials to high molecular weight polymers and may vary greatly in the nature of their backbone and substituent groups. For example, the backbone may be of any type and substituent groups thereon can be any group free of an active hydrogen atom which is reactive with an oxirane ring at room temperature. Illustrative of permissible substituent groups include halogens, ester groups, ethers, sulfonate groups, siloxane groups, nitro groups, phosphate groups, etc. The molecular weight of the epoxy-containing materials may vary from about 50 to 100,000 or more. Mixtures of various epoxy-containing materials can also be used in the compositions of this invention.

The epoxy compounds of the present invention may be cycloaliphatic epoxides. Examples of cycloaliphatic epoxides include diepoxides of cycloaliphatic esters of dicarboxylic acids such as bis(3,4-epoxycyclohexylmethyl)oxalate, bis(3,4-epoxycyclohexylmethyl)adipate, bis(3,4-epoxy-6-methylcyclohexylmethyl)adipate, bis(3,4-epoxycyclohexylmethyl)pimelate, and the like. Other suitable diepoxides of cycloaliphatic esters of dicarboxylic acids are described in, for example, U.S. Pat. No. 2,750,395, which is incorporated herein by reference. Other cycloaliphatic epoxides include 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylates such as 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylate; 3,4-epoxy-1-methylcyclohexylmethyl-3,4-epoxy-1-methylcyclohexane carboxylate; 6-methyl-3,4-epoxycyclohexylmethyl-6-methyl-3,4-epoxycyclohexane carboxylate; 3,4-epoxy-2-88methylcyclohexylmethyl-3,4-epoxy-2-methylcyclohexane carboxylate; 3,4-epoxy-3-

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methylcyclohexylmethyl-3,4-epoxy-3-methylcyclohexane carboxylate; 3,4-epoxy-5-methylcyclohexylmethyl-3,4-epoxy-5-methylcyclohexane carboxylate and the like. Other suitable 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylates are described in, for example, U.S. Pat. No. 2,890,194, which is incorporated herein by reference.

There are a host of commercially available epoxy-containing materials, commonly known as epoxy resins, which can be used as the epoxy compound in this invention. In particular, epoxy compounds which are readily available include octadecylene oxide, glycidylmethacrylate, diglycidyl ether of bisphenol A (e.g., those available under the trade designations EPON 828, EPON 1004 and EPON 1010 from Shell Chemical Co., DER-331, DER-332, and DER-334, from Dow Chemical Co.), vinylcyclohexene dioxide (e.g., ERL-4206 from Union Carbide Corp.), 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylate (e.g., ERL-4221 from Union Carbide Corp.), 3,4-epoxy-6-methylcyclohexylmethyl-3,4-epoxy-6-methylcyclohexene carboxylate (e.g., ERL-4201 from Union Carbide Corp.), bis(3,4-epoxy-6-methylcyclohexylmethyl) adipate (e.g., ERL-4289 from Union Carbide Corp.), bis(2,3-epoxycyclopentyl) ether (e.g., ERL-0400 from Union Carbide Corp.), aliphatic epoxy modified with polypropylene glycol (e.g., ERL-4050 and ERL-4052 from Union Carbide Corp.), dipentene dioxide (e.g., ERL-4269 from Union Carbide Corp.), epoxidized polybutadiene (e.g., OXIRON 2001 from FMC Corp.), silicone resin containing epoxy functionality, flame retardant epoxy resins (e.g., DER-580, a brominated bisphenol type epoxy resin available from Dow Chemical Co.), 1,4-butanediol diglycidyl ether of phenolformaldehyde novolak (e.g., DEN-431 and DEN-438 from Dow Chemical Co.), and resorcinol diglycidyl ether (e.g., KOPOXITE from Koppers Company, Inc.).

It is further contemplated that epoxy resins having a wide range of molecular weights and other characteristics are adaptable for use in accordance with the present invention. It is generally preferred, however, that the epoxy resins of the present invention possess an epoxy equivalent weight (EEW) of from about 180 to about 192. Furthermore, it is generally preferred that the epoxy resins have a viscosity of from about 11,000 to about 14,000 cps at about 25°C.

The epoxy resins suitable for use in the present invention may thus comprise one or more compounds, such as epoxy prepolymers, having more than one epoxide group per molecule available for reaction with the curing agent of the present invention. Such epoxy prepolymers include but are not limited to polyglycidyl ethers of polyvalent phenols, for example pyrocatechol, resorcinol, hydroquinone; 4,4'-dihydroxydiphenyl methane; 4,4'-dihydroxy-3,3'-dimethyldiphenyl methane; 4,4'-dihydroxydiphenyl dimethyl methane; 4,4'-dihydroxydiphenyl methyl methane; 4,4'-dihydroxydiphenyl cyclohexane; 4,4'-dihydroxy-3,3'-dimethyldiphenyl propane; 4,4'-dihydroxydiphenyl sulfone; or tris-(4-hydroxyphenyl)methane; polyglycidyl ethers of novalacs (i.e., reaction products of monohydric or polyhydric phenols with aldehydes, formaldehyde in particular, in the

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presence of acid catalysts); polyglycidyl ethers of diphenols obtained by esterifying 2 mols of the sodium salt of an aromatic hydrocarboxylic acid with 1 mol of a dihaloalkane or dihalogen dialkyl ether; and polyglycidyl ethers of polyphenols obtained by condensing phenols and long-chain halogen paraffins containing at least two halogen atoms.

Further epoxy-containing materials which are particularly useful in the practice of this invention include glycidyl ether monomers as disclosed in U.S. Patent No. 5,385,990, which is incorporated herein by reference.

Although it is contemplated that one or more epoxy compounds as described above may be adaptable for use within the scope of the present invention, in highly preferred embodiments the epoxy compound comprises in major proportion, and even more preferably is comprised of at least about 80% by weight of difunctional bisphenol A/epichlorohydrin derived liquid epoxy resin having a epoxide equivalent weight of from about 180 to about 200, and even more preferably from about 185 to about 195. As used herein, epoxide equivalent weight means the grams of epoxy compound or resin containing one gram equivalent of epoxide, as determined by the Shell Analytical Method HC-427D-89 (perchloric acid method). Such preferred epoxy also preferably has a viscosity at 25°C of from about 100 to about 175 poise, and even more preferably from about 110 to about 150 poise, as measured by Shell Analytical Method HC-397A-87. Such a preferred epoxy resin which satisfy these requirements is available from Shell Resins under the trade designation EPON Resin 828.

In preferred embodiments, reactive diluents are incorporated into the epoxy resin to, inter alia, control the flow characteristics of the adhesive composition. Suitable reactive diluents preferably have at least one reactive terminal end portion (including an epoxy group), and even more preferably also have a saturated or unsaturated cyclic backbone. Preferred reactive terminal ether portions include glycidyl ether and vinyl ether. Examples of suitable reactive diluents include 1,4-cyclohexane-dimethanol, the diglycidyl ether of resorcinol, diglycidyl ether of cyclohexane dimethanol, diglycidyl ether of neopentyl glycol, triglycidyl ether of trimethylolpropane dipentene, and the divinyl ether of cyclohexanedimethanol. Commercially available reactive diluents include "WC-68" from Rhone Poulenc, Epodil 741, 749 and 757 from Air Products and Chemicals Incorporated (Allentown, PA), Rapicure, a divinyl ether of cyclohexanedimethanol available from Allied-Signal Corp. of Morristown, N.J, and glycidyl neodeconate sold under the tradename CADURA E-10 by Shell Oil. Highly preferred for use in the epoxy resin as a reactive diluent is the 1,4 cyclohexane-dimethanol diglycidyl ether sold by Air Products under the trade designation Epodil 757.

In preferred embodiments of the present invention, the epoxy comprises, and preferably consists essentially of, epoxy resin and reactive diluent. In such embodiments, it is preferred that the epoxy resin:reactive diluent weight ratio is from about 65:10 to about 90:0.5, and even more preferably 70:5 to about 80:0.5. In highly preferred embodiments,

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the epoxy resin comprises about 97% by weight of epoxy (preferably EPON 828) and about 3% by weight of reactive diluent (preferably 1,4 cyclohexane-dimethanol diglycidyl ether).

The Amine Curing Agent

Although it is contemplated that, in general, the amine hardener may be any substance known as an amine-type curing agent for epoxy resins, it is generally preferred that the curing agent comprise, and preferably consist essentially of, a mixture of aliphatic and tertiary amines.

In certain preferred embodiments, the amine(s) include, on average, greater than two hydrogen atoms active towards the epoxide resin. The amines can contain one or more primary, secondary or tertiary nitrogen atoms.

Examples of suitable amines are aliphatic, cycloaliphatic, aromatic and heterocyclic amines, such as bis-(4-aminophenyl)-methane, aniline/formaldehyde resins, benzylamine, octylamine, propane-1,3-diamine, 2,2-dimethyl-1,3-propanediamine (neopentanediamine), hexamethylenediamine, diethylenetriamine, bis(3-aminopropyl)-amine, N,N-bis(3-aminopropyl)-methylamine, triethylenetetraamine, tetraethylenepentaamine, pentaethylenhexaamine, 2,2,4-trimethylhexane-1,6-diamine, m-xylylenediamine, 1,2-and 1,4-diaminocyclohexane, bis-(4-aminocyclohexyl)-methane, bis-(4-amino-3-methylcyclohexyl)-methane, 2,2-bis-(4-aminocyclohexyl)propane and 3-aminomethyl-3,5,5-trimethylcyclohexylamine (isophoronediamine), polyaminoimidazolines and polyaminoamides, for example those obtained from aliphatic polyamines and dimerized or trimerized fatty acids. Suitable amines also include the polyoxyalkyleneamines, known as Jeffamines, made by Huntsman Chemical, for example the Jeffamines EDR 148, D 230, D 400, or T 403, xylene diamine and bis(aminocyclohexyl) methane. Other curing agents include, for example, polyamines and polyamide curing agents.

For preferred embodiments in which the curing agent comprises a combination of aliphatic and tertiary amines, it is preferred that the aliphatic:tertiary weight ratio is from about 20:9 to about 35:0.5, more preferably from about 24:3 to about 30:5, and even more preferably from about 15:1 to about 25:1. In such embodiments, it is also generally preferred that the aliphatic amine compound(s) have a glass transition temperature of from about 125°F to about 140°F, more preferably from about 130°F to about 135°F, and even more preferably of about 133°F, and an amine value of at least about 350 mg KOH/g, more preferably of from about 350 mg KOH/g to about 800 mg KOH/g, and even more preferably from about 450 mg KOH/g to about 700 mg KOH/g. It is also generally preferred that the tertiary amine have a amine value of from about 400 mg KOH/g to about 800 mg KOH/g, more preferably from about 500 mg KOH/g to about 700 mg KOH/g, and even more preferably from about 600 mg KOH/g to about 650 mg KOH/g.

In highly preferred embodiments the aliphatic amines of the present invention comprise a first aliphatic amine having a glass transition temperature of from about 130°F

to about 135°F and an amine value of from about 400 mg KOH/g to about 500 mg KOH/g, and a second aliphatic amine having a glass transition temperature of from about 130°F to about 135°F and an amine value of at least about 550 mg KOH/g, and even more preferably of from about 550 mg KOH/g to about 700 mg KOH/g. As used herein, the "glass transition temperature" refers to the property as measured by ASTM D3418-82.

While it is contemplated that such embodiments may utilize such first and second aliphatic amines in a wide variety of relative amounts, it is generally preferred that the weight ratio of the first aliphatic amine to the second aliphatic amine is from about 1:1 to about 3:1, and more preferably of from about 1.5:1 to about 2.5:1 and even more preferably of from about 1.8:1 to about 2.1:1.

It is also contemplated that the amine curing agent may be present in a wide variety of concentration ranges within the broad scope of the present invention. Preferably, however, the present compositions comprise from about 5 wt% to about 50 wt% of amine curing agent, more preferably from about 10 wt% to about 40 wt% and even more preferably from about 10 wt% to about 20 wt%. Furthermore, applicants have found that unexpectedly superior results are generally achieved when the amine curing agent is present in the adhesive compositions in amounts of from about 20 pbwa to about 50 pbwa, and even more preferably about 30 pbwa to about 40 pbwa.

Fillers and Other Additives

The curable adhesive compositions of this invention may also contain other additives such as fillers, pigments, diluents and dyes or the like added to provide desired properties. Suitable filler for use in the compositions of this invention are mineral fillers. Illustrative examples include: talc, mica, titanium dioxide, lithopone, zinc oxide, zirconium, silica, silica aerogel, iron dioxide, diatomaceous earth, calcium carbonate, fumed silica, silazane treated silica, precipitated silica, glass fibers, magnesium oxide, chromic oxide, zirconium oxide, aluminum oxide, crushed quartz, calcined clay, asbestos, carbon, graphite, cork, cotton, synthetic fibers, to name but a few. Another filler is a fumed silica which also acts as a thixotropic agent and/or a compatibilizer, such as is sold under the trade name Cab-O-Sil TS-720. Another filler for use herein comprises a mixture of alumina, silica, and iron in the form of hollow spheres, which is available under the tradename Fillite 500 sold by the Fillite division of Boliden Intertrade Inc.

The preferred amount of filler will generally depend upon the particular type of adhesive being used, contemplated dispensing method, and anticipated use, among other factors. It is preferred however, that the weight ratio of the active components (that is, the epoxy(s) plus the curing agent amine(s)) to the filler is about 0.8:1 to about 2.2:1, and even more preferably from about 0.8:1 to about 1.5:1.

Generally, the other additives such as dyes, diluents, pigments, thixotropic agents, non-reactive diluents and the like will be added in conventional amounts.

TWO-PART EPOXY ADHESIVES

The preferred two-part epoxy adhesive comprises a first part (hereinafter referred to for convenience as "Part A") comprising epoxy and a second part (hereinafter referred to for convenience as "Part B") comprising the amine-based hardener or curing agent. Filler is preferably included in one or both of Part A and Part B. In such two part embodiments, Part A preferably comprises from about 65 to about 90 PBW of epoxy resin, from about 0.5 to about 9 PBW of reactive diluent and from about 9 to about 34 PBW of filler, and even more preferably from about 70 to about 80 PBW of epoxy resin, from about 0.5 to about 5 PBW of reactive diluent and from about 15 to about 25 PBW of filler.

Part B of the two-part adhesive compositions preferably contains the amine curing agent. The preferred curing agent comprises a combination of amines, particularly Mannich bases and aliphatic amines, available under the tradename Ancamine 1856 and Ancamine 2205 sold by Air Products and Chemicals Incorporated (Allentown, PA), and a catalyst, which is preferably a tertiary amine, particularly, 2,4,6-tri(dimethylaminomethyl)phenol, available under the tradename Ancamine K54 from Air Products and Chemicals Incorporated. According to certain preferred embodiments, Part B also comprises filler. Part B preferably comprises from about 20 to about 35 PBW of aliphatic amine, from about 0.5 to about 9 PBW of tertiary amine and from about 65 to about 80 PBW of filler, and even more preferably from about 24 to about 30 PBW of aliphatic amine, from about 0.5 to about 3 PBW of tertiary amine, and from about 65 to about 75 PBW of filler.

The amount of Part B in the present compositions is preferably about 100 parts by weight and/or by volume per 100 parts of part A.

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WHAT IS CLAIMED IS:

1. A curable adhesive composition for anchoring materials in or to concrete or masonry comprising:
at least one polymerizable epoxy compound; and
a curing agent comprising at least one aliphatic amine and at least one tertiary amine, said epoxy compound and said curing agent being present in amounts effective to produce an adhesive composition which passes ICBO Heat Creep Test (ICBO acceptance criteria AC58) at 110°F and which passes the ICBO Damp Hole Test at 75°F.
2. The curable adhesive composition of claim 1 wherein said polymerizable epoxy compound comprises at least one compound derived from the reaction of difunctional bisphenol-A and epichlorohydrin.
3. The curable adhesive composition of claim 1 wherein said polymerizable epoxy compound has an epoxide equivalent weight of from about 180 to about 200.
4. The curable adhesive composition of claim 3 wherein said polymerizable epoxy compound has an epoxide equivalent weight of from about 180 to about 190.
5. The curable adhesive composition of claim 1 wherein said epoxy is present in the composition in an amount of from about 20% by weight to about 40% by weight.
6. The curable adhesive composition of claim 5 wherein said polymerizable epoxy compound is present in the adhesive compositions in an amount of from about 70 parts by weight of active components (pbwa) to about 95 pbwa.
7. The curable adhesive composition of claim 1 wherein said curing agent comprises a combination of aliphatic and tertiary amines.
8. The curable adhesive composition of claim 7 wherein said curing agent consists essentially of a combination of aliphatic and tertiary amines.
9. The curable adhesive composition of claim 7 having an epoxy:amine weight ratio of from about 0.5:1 to about 10:1.
10. The curable adhesive composition of claim 7 having an epoxy:amine weight ratio of from about 0.7:1 to about 2:1.
11. A curable adhesive composition for anchoring materials in or to concrete or masonry comprising:
from about 20% by weight to about 40% by weight of at least one epoxy compound derived from the reaction of difunctional bisphenol-A and epichlorohydrin and having an epoxide equivalent weight of from about 180 to about 190; and
a curing agent comprising amine curing agent in an amount sufficient to produce an epoxy:amine weight ratio of from about 0.8:1 to about 1:1, said amine

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- curing agent comprising at least one aliphatic amine and at least one tertiary amine.
12. The curable adhesive of claim 11 wherein said amine curing agent consists essentially of at least one aliphatic amine and at least one tertiary amine. .
 13. The curable adhesive composition of claim 12 wherein said at least one aliphatic amine comprises a first aliphatic amine having a glass transition temperature of from about 130°F to about 135°F and an amine value of from about 400 mg KOH/g to about 500 mg KOH/g and a second aliphatic amine having a glass transition temperature of from about 130°F to about 135°F and an amine value of at least about 550 mg KOH/g.
 14. The curable adhesive composition of claim 13 wherein said second aliphatic amine has an amine value of from about 550 mg KOH/g to about 700 mg KOH/g.
 15. The adhesive composition of claim 14 further comprising reactive diluent for said polymerizable epoxy compound.
 16. The adhesive composition of claim 15 wherein the epoxy:reactive diluent weight ratio is from about 65:10 to about 90:0.5.
 17. The adhesive composition of claim 15 wherein the epoxy:reactive diluent weight ratio of from about 70:5 to about 80:0.5.
 18. The adhesive composition of claim 1 having a cure time of no greater than about 2.5 hours.
 19. The adhesive composition of claim 1 having a cure time of no greater than about 2.5 hours.

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INTERNATIONAL SEARCH REPORT

International application No.

PCT/US02/41424

A. CLASSIFICATION OF SUBJECT MATTER

IPC(7) : C08G 59/50, 59/54, 59/56; C08L 63/00, 63/02

US CL : 525/423, 523; 528/120, 121, 122, 123

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 525/423, 523; 528/120, 121, 122, 123

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
CAS Online: Files REG and CAPLUS**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X — Y	US 6,291,555 B1 (SURJAN et al.) 18 September 2001 (18.09.01), column 2, lines 14-16; column 4, lines 28-31 and 40-48; and column 5, lines 21-25 and 46-55.	1-12, 18 and 19 ----- 13-17
T —	US 6,403,678 B1 (SURJAN et al.) 11 June 2002 (11.06.02), column 2, lines 11-12; column 5, lines 24-27 and 36-44; and column 6, lines 17-22 and 42-55.	1-12, 18 and 19
T —	US 6,420,458 B1 (SURJAN et al.) 16 July 2002 (16.07.02), column 2, lines 11-13 and 19-29; column 3, lines 44-47 and 56-64; column 4, lines 36-40 and 61-67; and column 5, lines 1-7.	1-12, 18 and 19
X — Y	US 5,962,602 A (HARTMAN et al.) 05 October 1999 (05.10.99), column 2, lines 52-60 and column 3, lines 14-32 and 47-66.	1-19 ----- 1-19
Y —	JP 2001-240837 A2 (SANYO CHEMICAL IND.) 09 April 2001 (09.04.01), abstract AB and IT term for 90-72-2, DMP 30.	1-12, 18 and 19

☐ Further documents are listed in the continuation of Box C.☐ See patent family annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T"

later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X"

document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y"

document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&"

document member of the same patent family

Date of the actual completion of the international search

01 April 2003 (01.04.2003)

Date of mailing of the international search report

25 APR 2003

Name and mailing address of the ISA/US

Commissioner of Patents and Trademarks
Box PCT
Washington, D.C. 20231

Facsimile No. (703)305-3230

Authorized officer

Robert Sellers

DEBORAH THOMAS
PARALEGAL SPECIALIST

Telephone No. (703) 308-1193

Form PCT/ISA/210 (second sheet) (July 1998)

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INTERNATIONAL SEARCH REPORT

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Continuation of Item 4 of the first sheet:

The title does not meet the requirements of PCT Rule 4.3 because it does not precisely indicate the subject matter. The following new title is provided.

NEW TITLE: ADHESIVE OF EPOXY COMPOUND, ALIPHATIC AMINE AND TERTIARY AMINE

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